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| https://encrypted-tbn0.gstatic.com/images?q=tbn:ANd9GcQKqu6GRuBsVmXccvhYUzX5vT9FK30R7cesCqY2p8G7nMjJXhrUxAHomework:   1. If an atom has 12 protons and 10 neutrons, it has a mass of \_\_\_\_\_\_ and it is the element \_\_\_\_\_\_\_\_\_\_. 2. If an atom has 1 proton and 0 neutrons, and 2 electrons, what is the highLow symbol of the atom? \_\_\_\_\_\_\_\_ 3. If an atom is helium how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_ 4. If an atom has 10 protons what element is it? \_\_\_\_\_\_\_\_\_\_\_\_\_\_ 5. If the atom’s mass number is 18 how many P and N, total does it have? 6. If an atom has 12 protons and it has 25 mass, how many neutrons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 7. If an atom has 20 mass number and has 9 neutrons, how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 8. If an atom has 1 proton and 0 neutrons what element is it? \_\_\_\_\_\_\_\_ 9. If an atom has 22 protons and 20 neutrons, and 20 electrons, write the highLow symbol with charge \_\_\_\_\_\_\_\_\_\_. 10. If an atom has 8 proton and 11 neutrons what is the mass of the atom? \_\_\_\_\_\_\_\_ 11. If an atom is magnesium how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_ 12. If an atom has 15 protons what element is it? \_\_\_\_\_\_\_\_\_\_\_\_\_\_ 13. If an atom mass number is 28 amu's and there are 12 neutrons, what element is it? 14. If an atom has 6 protons and it has 10 mass number, how many neutrons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 15. If an atom has 1 mass number and has zero neutrons, how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 16. If an atom has 92 protons, 91 electrons, and 146 neutrons write the symbol, with highLow numbers and charge \_\_\_\_\_\_\_\_ | https://encrypted-tbn0.gstatic.com/images?q=tbn:ANd9GcQKqu6GRuBsVmXccvhYUzX5vT9FK30R7cesCqY2p8G7nMjJXhrUxAHomework:   1. If an atom has 12 protons and 10 neutrons, it has a mass of \_\_\_\_\_\_ and it is the element \_\_\_\_\_\_\_\_\_\_. 2. If an atom has 1 proton and 0 neutrons, and 2 electrons, what is the highLow symbol of the atom? \_\_\_\_\_\_\_\_ 3. If an atom is helium how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_ 4. If an atom has 10 protons what element is it? \_\_\_\_\_\_\_\_\_\_\_\_\_\_ 5. If the atom’s mass number is 18 how many P and N, total does it have? 6. If an atom has 12 protons and it has 25 mass, how many neutrons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 7. If an atom has 20 mass number and has 9 neutrons, how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 8. If an atom has 1 proton and 0 neutrons what element is it? \_\_\_\_\_\_\_\_ 9. If an atom has 22 protons and 20 neutrons, and 20 electrons, write the highLow symbol with charge \_\_\_\_\_\_\_\_\_\_. 10. If an atom has 8 proton and 11 neutrons what is the mass of the atom? \_\_\_\_\_\_\_\_ 11. If an atom is magnesium how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_ 12. If an atom has 15 protons what element is it? \_\_\_\_\_\_\_\_\_\_\_\_\_\_ 13. If an atom mass number is 28 amu's and there are 12 neutrons, what element is it? 14. If an atom has 6 protons and it has 10 mass number, how many neutrons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 15. If an atom has 1 mass number and has zero neutrons, how many protons does it have? \_\_\_\_\_\_\_\_\_\_\_\_ 16. If an atom has 92 protons, 91 electrons, and 146 neutrons write the symbol, with highLow numbers and charge \_\_\_\_\_\_\_\_ |