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| Ionization Energy for removing all the electrons from BORON, one by oneBelow are graphs of the ionization energies for BORON. Two definitions to memorize:Electron Affinity is defined as the energy required to *add* an electron to an atom.Ionization energy is defined as the energy required to *remove* an electron from an atom. It is not the energy for adding an electron. The graphs here are for ionization energy only. |

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| Ionization Energy for removing all the electrons from SODIUM, one by one  |

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| Ionization Energy for removing all the electrons from Calcium, one by one |

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| Ionization Energy for removing all the electrons from \_\_\_\_\_\_\_\_ one by one |

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