|  |
| --- |
| Ionization Energy for removing all the electrons from BORON, one by one  Below are graphs of the ionization energies for BORON.  Two definitions to memorize:  Electron Affinity is defined as the energy required to *add* an electron to an atom.  Ionization energy is defined as the energy required to *remove* an electron from an atom. It is not the energy for adding an electron.  The graphs here are for ionization energy only. |

|  |
| --- |
| Ionization Energy for removing all the electrons from SODIUM, one by one |

|  |
| --- |
| Ionization Energy for removing all the electrons from Calcium, one by one |

|  |
| --- |
| Ionization Energy for removing all the electrons from \_\_\_\_\_\_\_\_ one by one |

|  |
| --- |
|  |

|  |
| --- |
|  |