

Homework tonight: Top Hat sheet

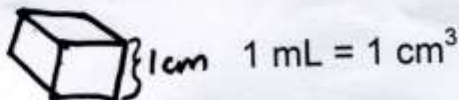
Turn in by 11:10: The graph

Purpose :

What is a "slope" GOOD FOR?

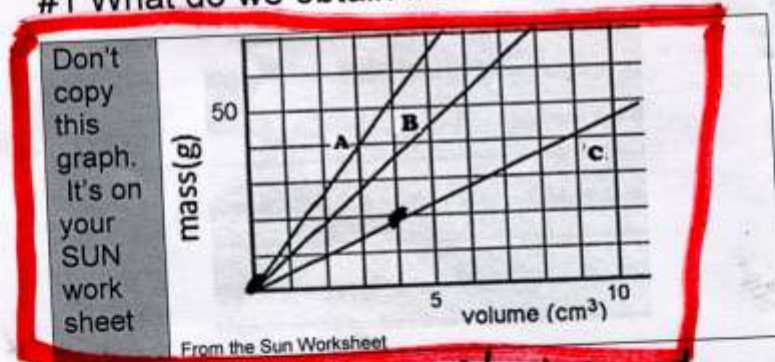
Warmup (copy this):

Milliliter (definition): a mL is the volume inside of a cube that is 1cm x 1cm x 1cm. In other words



This has ∞ sig figs.

#1 What do we obtain from a best fit line?



We obtain a slope with top and bottom units. Line C: $\frac{\Delta y}{\Delta x} = \frac{20 \text{ grams}}{4 \text{ cm}^3}$

#2 How do we put a slope into a math equation?

Useful:

$$\frac{1.2 \text{ grams}}{11.7 \text{ grams}} * \frac{1 \text{ mL}}{11.7 \text{ grams}} = 0.10 \text{ mL}$$

Not useful:

$$\frac{1.2 \text{ grams}}{11.7 \text{ grams}} * \frac{11.7 \text{ grams}}{1 \text{ mL}} = 10 \frac{\text{g}^2}{\text{mL}}$$

Sample Problem

"What would a coin weigh if it were made of Thorium? We know that the density of thorium is 11.7 grams / 1 mL"

$$0.3 \text{ mL} \times \left(\frac{1 \text{ mL}}{11.7 \text{ g}} \right) = \frac{\text{mL}^2}{\text{g}}$$

from the slope
of density

$$0.3 \text{ mL} \times \left(\frac{11.7 \text{ g}}{1 \text{ mL}} \right) = 3.2 \text{ grams}$$