

Applied Density Practice

EHS CA3MIs+ry

Mr. Genest



Name \_\_\_\_\_  
Date \_\_\_\_\_

Tutors! Adults! Help this young chemist by visiting <http://genest.weebly.com> with any smart phone

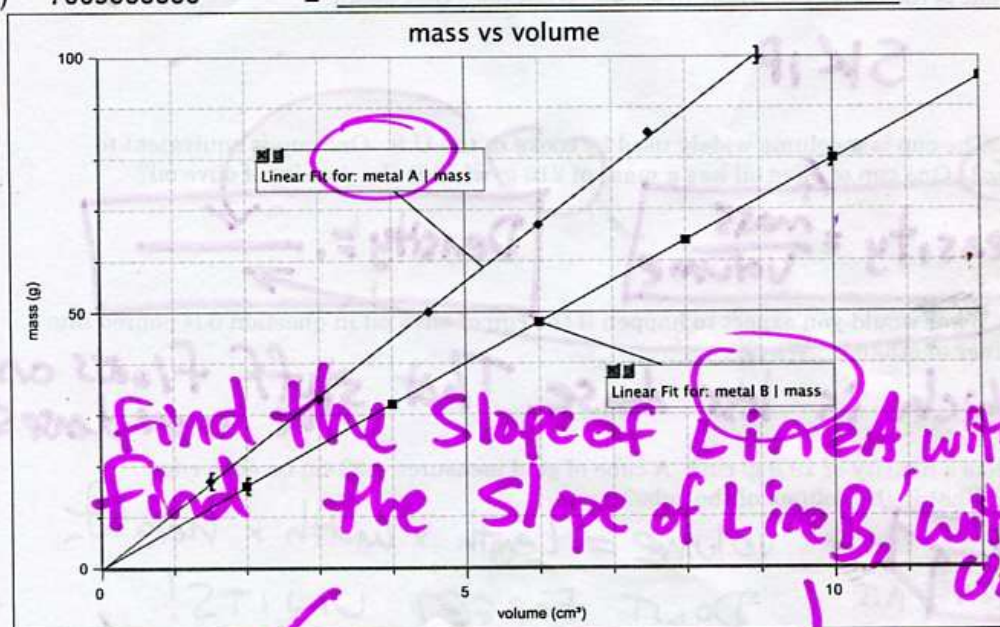
Write each number in scientific notation.

11 ) 529000000 = \_\_\_\_\_

12 ) 5.29 = \_\_\_\_\_

13 ) 982.05 = \_\_\_\_\_

14 ) 7009000000 = \_\_\_\_\_



1. Determine the density of each metal (hint: density is just the slope of grams per mL). Show all your work and include appropriate units.

Line A slope = ?

Line B slope = ?

2. From the graph, estimate

a. the mass of 8.0 cm<sup>3</sup> of metal A.

b. the volume of 70 g of metal B.

c. mark on the graph how you found the answers above.

touch line A at 8.0 cm<sup>3</sup>. Read the grams.  
touch line B at 70 g. Read the cm<sup>3</sup>

3. Use the density of B as a factor to determine the answer to 2b. Show the set-up including how the units cancel.

$70 \text{ grams} \times \left( \frac{\text{cm}^3}{\text{g}} \right) =$

*use slope of B with units*

4. Ethanol has a density of  $0.789 \text{ g/cm}^3$ .
- a. What is the mass of  $225 \text{ cm}^3$  of ethanol?

$225 \text{ cm}^3 \times \left( \frac{0.789 \text{ g}}{1 \text{ cm}^3} \right) =$

- b. What is the volume of  $75.0 \text{ g}$  of ethanol?

$75.0 \text{ g} \times \left( \frac{1 \text{ cm}^3}{0.789 \text{ g}} \right) =$

5. What is the density of water in  $\text{g/mL}$ ? What does that mean?

SKIP

6. The cup is a volume widely used by cooks in the U.S. One cup is equivalent to  $237 \text{ cm}^3$ . One cup of olive oil has a mass of  $216 \text{ g}$ , what is the density of olive oil?

$\text{Density} = \frac{\text{mass}}{\text{Volume}}$

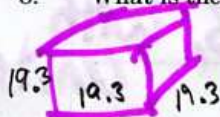
$\text{Density} = \frac{216 \text{ g}}{237 \text{ cm}^3}$

7. What would you expect to happen if the cup of olive oil in question 6 is poured into a container of ethanol? Why?

*Find which is less dense. That stuff floats on the more dense stuff.*

Gold has a density of  $19.3 \text{ g/cm}^3$ . A cube of gold measures  $4.23 \text{ cm}$  on each edge:

8. What is the volume of the cube?



$\text{Volume} = \text{Length} \times \text{width} \times \text{height}$

**DON'T FORGET UNITS!**

9. What is its mass? How many significant figures should you include in your answer and why?

10. A standard backpack is approximately  $30 \text{ cm} \times 30 \text{ cm} \times 40 \text{ cm}$ . Suppose you find a hoard of pure gold while treasure hunting in the wilderness. How much mass would your backpack hold if you filled it with the gold? An average student has a mass of  $70 \text{ kg}$ . How do these values compare?