

## TODAY

- 1) notebooks
- 2) a demonstration
- 3) your group presents a whiteboard up front


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Purpose: What are chemical changes?

### Warm up:

PHYSICAL CHANGES are things like boiling, freezing, and dissolving. These are called phase changes or physical changes.

Chemical Changes are any change that creates a new substance with new properties.

important part! 



Name \_\_\_\_\_

Period \_\_\_\_\_

In a physical change, the original substance still exists. It has only changed in form. In chemical change, a new substance is produced. Energy changes always accompany chemical changes.

Classify the following as being a physical or chemical change.

- Sodium hydroxide dissolves in water. PHYSICAL  
(JUST A PHASE CHANGE)
- Hydrochloric acid reacts with potassium hydroxide to produce a salt, water and heat. CHEMICAL.
- A pellet of sodium is sliced in two. \_\_\_\_\_
- Water is heated and changed to steam. \_\_\_\_\_
- Potassium chlorate decomposes to potassium chloride and oxygen gas.  
\_\_\_\_\_
- Iron rusts. \_\_\_\_\_
- When placed in  $H_2O$ , a sodium pellet catches on fire as hydrogen gas is liberated and sodium hydroxide forms. \_\_\_\_\_
- Evaporation \_\_\_\_\_
- Ice melting \_\_\_\_\_
- Milk sours. \_\_\_\_\_
- Sugar dissolves in water. \_\_\_\_\_
- Wood rotting \_\_\_\_\_
- Pancakes cooking on a griddle \_\_\_\_\_
- Grass growing in a lawn \_\_\_\_\_
- A tire is inflated with air. \_\_\_\_\_
- Food is digested in the stomach. \_\_\_\_\_
- Water is absorbed by a paper towel. \_\_\_\_\_