

I'm here at lunch every day.

Come for just 5 minutes, it helps.

**Purpose:**

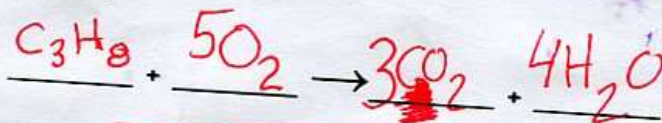
What's the two-finger method for balancing a reaction?

**WARMUP:**

Sign up for a problem, grab a whiteboard, and be ready to present at 11:16

#1

Write a balanced reaction for the combustion of  $C_3H_8$ .

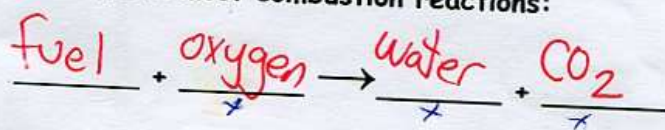


3	C	3
8	H	8
2	O	10

check your work with a scoreboard

#2

Memorize the FOUR reactants and products that are in most combustion reactions:



check your work with a scoreboard

#3

Write a balanced reaction for the combustion of ~~CH<sub>4</sub>O~~ CH<sub>4</sub>.

