|  |
| --- |
|  |

|  |
| --- |
|  |

|  |
| --- |
|  |

|  |  |  |
| --- | --- | --- |
| *Warmup sheet: Keep and study*CλeMis+ry eagenest@madison.k12.wi.us | Air Jordan logo | Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Period\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ |

|  |  |
| --- | --- |
|  | Using your numbers from yesterday’s warmup sheet, calculate the mass, in grams, of the glycerine on the table topWrite a balanced reaction for the combustion of this substanceCalculate how many moles of carbon dioxide would form if you combusted the quantity of glycerine taped to the tabletop. |

|  |
| --- |
| Key to understanding the cartoons on this sheet: |
| 1 chlorine atom | 1 hydrogen atom | 1 oxygen atom | 1 nitrogen atom | 1 carbon atom |
|  |  |  |  |  |