

TODAY: ① H.W. CHECK ② LAB, DO #1 - #5 ③ Take Quiz
 ④ DO LAB STEPS #6, #7

WARMUP:

The balanced equation for the combustion of propane (C_3H_8) is shown below. How many moles of oxygen are needed to completely burn 8 moles of propane?



B	8 mol	?	ZERO	ZERO
C	-8 mol	-40	+	+
A	Zero			

Answer: 40 moles O_2