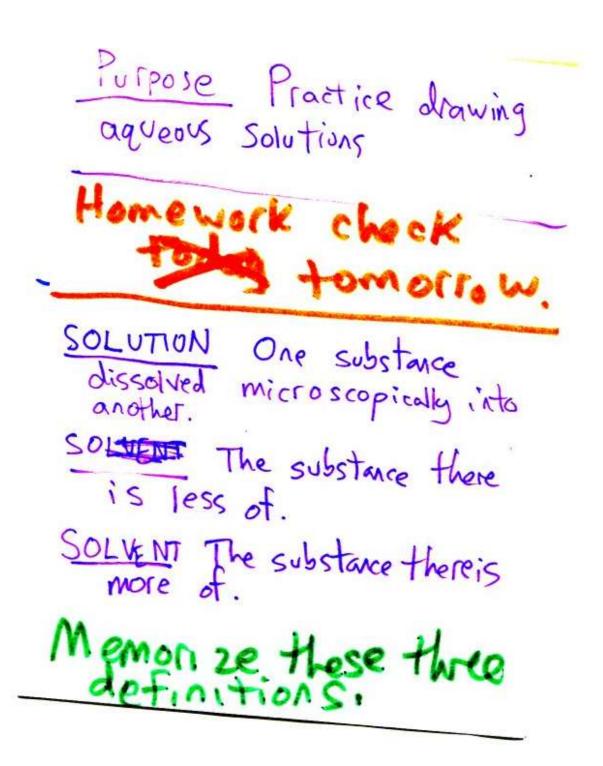
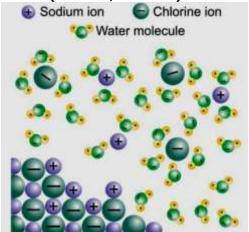
No homework tonight.

Make sure you did Tuesday's night's homework, because we will check it Thursday in class.



1. In this picture the solvent is (NaCl / water) and the solute is (NaCl / water).



2. Write the name of your two substances for your poster here:

First Substance	
Second Substance	re

3. Write a balanced reaction for each of these going from solid to aqueous ions:

First Substance		 +	
Second Substance	<b>→</b>	 +	

- 4. Draw a box in your notebook and show 20 formula units of your first substance into all of the solid
- 5. Draw another box and show 20 formula units of your second substance as aqueous particles in the box. Show some H<sub>2</sub>O It's okay to make single circles for polyatomics:









No homework tonight.

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